

## 14.1

## HYDROGEN IONS AND ACIDITY

## Section Overview

## Objectives

1. Identify conjugate acid-base pairs of ions, acids, or hydroxides and an appropriate equilibrium.
2. Calculate hydroxide and hydronium ion concentrations and pH, and vice versa, for strong acids and bases.
3. Calculate the pH of weak acid solutions.

## Vocabulary

- |                   |                             |                         |
|-------------------|-----------------------------|-------------------------|
| 1. autoionization | 2. conjugate acid-base pair | 3. strong acid/base     |
| 4. hydronium ion  | 5. weak acid/base           | 6. equilibrium constant |
|                   |                             | 7. pH                   |

## Key Equations

- |  |  |
|--|--|
| 1. $\text{H}_2\text{O} + \text{H}_2\text{O} \rightleftharpoons \text{H}_3\text{O}^+ + \text{OH}^-$ | 2. $\text{pH} = -\log[\text{H}_3\text{O}^+]$                                     |
| 3. $\text{pH} + \text{pOH} = \text{p}K_w$  | 4. $\text{pH} = \text{p}K_a + \log\left(\frac{[\text{A}^-]}{[\text{HA}]}\right)$ |

## Part A: Completion

Use the appropriate number of significant figures in the answers to the following questions. Show calculations for all problems. Answers should be written in the space provided.

1. What products are formed when hydrochloric acid reacts with sodium hydroxide? Write the balanced chemical equation for this reaction and identify the products as strong or weak acids/bases.
2. Hydrochloric acid has a concentration of 0.010 M. Calculate the pH of this solution.
3. Calculate the pH of a 0.010 M solution of hydrochloric acid.
4. Calculate the pH of a 0.010 M solution of sodium hydroxide.
5. Calculate the pH of a 0.010 M solution of acetic acid.
6. Calculate the pH of a 0.010 M solution of sodium acetate.
7. Calculate the pH of a 0.010 M solution of acetic acid.
8. Calculate the pH of a 0.010 M solution of sodium acetate.

9.  $\text{HCl} + \text{NaOH} \rightarrow \text{NaCl} + \text{H}_2\text{O}$
10.  $\text{HCl} + \text{NaOH} \rightarrow \text{NaCl} + \text{H}_2\text{O}$
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# 19 2 Hydrogen Ions And Acidity Answers

**Lingsheng Yao**



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**The Industrial Arts Index**, 1930 *Annual Report of the Dairy and Food Commissioner of Wisconsin* Wisconsin Dairy and Food Commission, 1917

**SAT Two, Biology and Biology E/M** Maurice Bleifeld, 1998 An overview of the SAT II biology exams with a review of test taking strategies is followed by a full length diagnostic test review chapters covering 11 biology topics and five complete practice tests each with an answer key a self evaluation chart and explanations of answers

*General Chemistry with Qualitative Analysis* Ralph H. Petrucci, Robert K. Wismer, 1983

**Technical Association Papers** Technical Association of the Pulp and Paper Industry, 1929

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