

- vii. **Name two non-metals which are both brittle and non-ductile**  
**Ans.** Graphite and iodine are two non-metals which are brittle and non-ductile in nature.
- viii. **Name the most abundant non-metal in the earth's crust**  
**Ans.** The most abundant non-metal in the earth's crust is oxygen. It is 47 % of earth's crust.
- ix. **Give the non-metallic trend in halogens**  
**Ans.** The non-metallic trend in halogens decreases from top to bottom because of increasing atomic sizes.
- x. **Why do the non-metals accept electrons readily?**  
**Ans.** Non-metals accept electrons readily because they are more electronegative and are usually electron deficient in nature. So they form anions by gaining electrons.



- xi. **Why non-metals do not react with dilute acids while metals do react?**  
**Ans.** Non-metals do not react with dilute acids because non-metals are itself electron acceptors while metals react readily because they lose electrons readily.
- xii. **How can we distinguish a metal from a non-metal by simple physical methods?**  
**Ans.**

Metals	Non-metals
i. Metals are good conductor of heat and electricity. ii. Metals possess high melting and boiling points. iii. They are lustrous.	i. Non metals are bad conductor of heat and electricity. ii. Non metals, possess low melting and boiling points. iii. They are dull.

- xiii. **How we can distinguish a substance is metal or non-metals with the help of an acid?**  
**Ans.**

Metal	Non metal
Metals react with dilute acids easily because metals are itself electron donors $\text{Zn} + 2\text{HCl} \longrightarrow \text{ZnCl}_2 + \text{H}_2$ $\text{Zn} + \text{H}_2\text{SO}_4 \longrightarrow \text{ZnSO}_4 + \text{H}_2$	Non-metals or usually do not react with dilute acids because non-metals are itself electron accepters.

- xiv. **Why is HF a weak acid?**  
**Ans.** HF is a weak acid because it does not release its proton easily due to presence of strong hydrogen bonding and it ionizes to a small extent in aqueous solution.  
**Fig:** H- bonding in HF molecules

# Quiz Yourself Chapter 8 Introductcion To Chemistry

**Daniel F McAuley**



**Quiz Yourself Chapter 8 Introduction To Chemistry:**

This book delves into Quiz Yourself Chapter 8 Introduction To Chemistry. Quiz Yourself Chapter 8 Introduction To Chemistry is a vital topic that needs to be grasped by everyone, from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Quiz Yourself Chapter 8 Introduction To Chemistry, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Quiz Yourself Chapter 8 Introduction To Chemistry
- Chapter 2: Essential Elements of Quiz Yourself Chapter 8 Introduction To Chemistry
- Chapter 3: Quiz Yourself Chapter 8 Introduction To Chemistry in Everyday Life
- Chapter 4: Quiz Yourself Chapter 8 Introduction To Chemistry in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Quiz Yourself Chapter 8 Introduction To Chemistry. The first chapter will explore what Quiz Yourself Chapter 8 Introduction To Chemistry is, why Quiz Yourself Chapter 8 Introduction To Chemistry is vital, and how to effectively learn about Quiz Yourself Chapter 8 Introduction To Chemistry.

3. In chapter 2, the author will delve into the foundational concepts of Quiz Yourself Chapter 8 Introduction To Chemistry. This chapter will elucidate the essential principles that must be understood to grasp Quiz Yourself Chapter 8 Introduction To Chemistry in its entirety.

4. In chapter 3, this book will examine the practical applications of Quiz Yourself Chapter 8 Introduction To Chemistry in daily life. This chapter will showcase real-world examples of how Quiz Yourself Chapter 8 Introduction To Chemistry can be effectively utilized in everyday scenarios.

5. In chapter 4, the author will scrutinize the relevance of Quiz Yourself Chapter 8 Introduction To Chemistry in specific contexts. The fourth chapter will explore how Quiz Yourself Chapter 8 Introduction To Chemistry is applied in specialized fields, such as education, business, and technology.

6. In chapter 5, the author will draw a conclusion about Quiz Yourself Chapter 8 Introduction To Chemistry. The final chapter will summarize the key points that have been discussed throughout the book.

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