

# Voltaic Cells

Label the parts for the following cell:



By connecting to the voltmeter, describe a spontaneous redox reaction. (10 marks)

- a. metals \_\_\_\_\_
- b. metal ions \_\_\_\_\_
- c. metals \_\_\_\_\_
- d. metal ions \_\_\_\_\_
- e. electron movement flow \_\_\_\_\_
- f. displacement overall movement \_\_\_\_\_

- g. half cell reduction equation \_\_\_\_\_
- h. Zn \_\_\_\_\_

i. Write overall cell equation with phases.  $\text{cell} = \text{anode} | \text{anode ion} || \text{cathode ion} | \text{cathode}$  represent by net cell equation with half cell notation for anode & cathode

\_\_\_\_\_

# Voltaic Cells Study Guide

**Tobias Bleicker**



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