

18.4 Acid-Base Chemistry: Theory

Multiple-Choice Questions

1. Which combination of quantities are dimensional quantities? (correct)
- (A) pressure, concentration, pH, amount of substance
 (B) volume, molarity, pH, amount of substance
 (C) volume, molarity, pH, concentration of ions
 (D) amount of substance, pH, concentration of ions, concentration of substance

(A) $\text{pH} = \frac{[\text{H}^+]}{10^{-7}}$ (B) $\text{pH} = \frac{[\text{H}^+]}{10^{-14}}$

(C) $\text{pH} = \frac{[\text{H}^+]}{10^{-7}}$ (D) $\text{pH} = \frac{[\text{H}^+]}{10^{-14}}$ (E) $\text{pH} = \frac{[\text{H}^+]}{10^{-7}}$ (F) $\text{pH} = \frac{[\text{H}^+]}{10^{-14}}$

2. An acid-base indicator has a pK_a of 4.5. In a solution of the indicator, the ratio of the concentration of the acid form to the concentration of the base form is 10:1. What is the pH of the solution? (correct)

(A) $\text{pH} = 4.5$ (B) $\text{pH} = 5.5$ (C) $\text{pH} = 6.5$ (D) $\text{pH} = 7.5$ (E) $\text{pH} = 8.5$

(A) $\text{pH} = 4.5$ (B) $\text{pH} = 5.5$ (C) $\text{pH} = 6.5$ (D) $\text{pH} = 7.5$ (E) $\text{pH} = 8.5$

3. An acid-base indicator has a pK_a of 4.5. In a solution of the indicator, the ratio of the concentration of the acid form to the concentration of the base form is 10:1. What is the pH of the solution? (correct)

(A) $\text{pH} = 4.5$ (B) $\text{pH} = 5.5$ (C) $\text{pH} = 6.5$ (D) $\text{pH} = 7.5$ (E) $\text{pH} = 8.5$

4. Which of the following is a buffer solution? (correct)

(A) HCl and NaCl (B) HCl and NaOH

(A) HCl (B) NaCl (C) HCl (D) NaCl

5. A solution of an acid and its conjugate base is prepared. The pH of the solution is 4.5. The pK_a of the acid is 4.5. What is the ratio of the concentration of the acid to the concentration of the base? (correct)

(A) 10:1 (B) 1:10 (C) 1:100 (D) 1:1000 (E) 1:10000

(A) $\text{pH} = 4.5$ (B) $\text{pH} = 5.5$ (C) $\text{pH} = 6.5$ (D) $\text{pH} = 7.5$ (E) $\text{pH} = 8.5$

6. An acid-base indicator has a pK_a of 4.5. In a solution of the indicator, the ratio of the concentration of the acid form to the concentration of the base form is 10:1. What is the pH of the solution? (correct)

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Ying-Ying Zheng



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Table of Contents Multiple Choice Stoichiometry Test

1. Understanding the eBook Multiple Choice Stoichiometry Test
 - The Rise of Digital Reading Multiple Choice Stoichiometry Test
 - Advantages of eBooks Over Traditional Books
2. Identifying Multiple Choice Stoichiometry Test
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Multiple Choice Stoichiometry Test
 - User-Friendly Interface
4. Exploring eBook Recommendations from Multiple Choice Stoichiometry Test

- Personalized Recommendations
 - Multiple Choice Stoichiometry Test User Reviews and Ratings
 - Multiple Choice Stoichiometry Test and Bestseller Lists
5. Accessing Multiple Choice Stoichiometry Test Free and Paid eBooks
 - Multiple Choice Stoichiometry Test Public Domain eBooks
 - Multiple Choice Stoichiometry Test eBook Subscription Services
 - Multiple Choice Stoichiometry Test Budget-Friendly Options
 6. Navigating Multiple Choice Stoichiometry Test eBook Formats
 - ePub, PDF, MOBI, and More
 - Multiple Choice Stoichiometry Test Compatibility with Devices
 - Multiple Choice Stoichiometry Test Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Multiple Choice Stoichiometry Test
 - Highlighting and Note-Taking Multiple Choice Stoichiometry Test
 - Interactive Elements Multiple Choice Stoichiometry Test
 8. Staying Engaged with Multiple Choice Stoichiometry Test
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Multiple Choice Stoichiometry Test
 9. Balancing eBooks and Physical Books Multiple Choice Stoichiometry Test
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Multiple Choice Stoichiometry Test
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Multiple Choice Stoichiometry Test
 - Setting Reading Goals Multiple Choice Stoichiometry Test
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Multiple Choice Stoichiometry Test

- Fact-Checking eBook Content of Multiple Choice Stoichiometry Test
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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